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(54) Title: PROCESS FOR THE PRODUCTION OF AMORPHOUS ATORVASTATIN CALCIUM

(57) Abstract: A process for the preparation of amorphous atorvastatin calcium and hydrates thereof which comprises: (a) dissolving crystalline atorvastatin calcium in a non-hydroxylic solvent; (b) adding a non-polar hydrocarbon anti-solvent or adding the dissolved atorvastatin to the non-polar anti-solvent to precipitate out atorvastatin calcium; and (c) removing the solvent by filtration to afford amorphous atorvastatin calcium.

# PROCESS FOR THE PRODUCTION OF AMORPHOUS ATORVASTATIN CALCIUM

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#### FIELD OF THE INVENTION

The present invention relates to a process for the production of amorphous atorvastatin calcium.

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## **BACKGROUND OF THE INVENTION**

Atorvastatin is chemically [R-(R\*,R\*)]-2-(4-fluoro-phenyl)-β dihydroxy-5-(1-methylethyl)-3-phenyl-4-[(phenylamino) carbonyl]-1H pyrrole-1-heptanoic acid. Atorvastatin calcium, a synthetic HMG-CoA reductase inhibitor, is used for the treatment of hyperlipidemia and hypercholesterolemia, both of which are risk factors for arteriosclerosis and coronary heart disease. Open dihydroxy carboxylic acid, lactone and various salt forms of atorvastatin have been synthesized.

United States Patent 5,273,995, describes that R-form of the ring opened acid form has surprising inhibition of the biosynthesis of cholesterol. Atorvastatin in its calcium salt form, i.e.  $[R-(R^*,R^*)]-2-(4-fluoro-phenyl)-\beta$ ,  $\delta$ -dihydroxy-5-(1-methylethyl)-3-phenyl-4- $[(phenylamino) \quad carbomyl]-1H$ -pyrrole-1-heptanoic acid calcium salt (2:1) having Formula 1:

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is more suited to formulations and has been recommended as a drug.

United States patents 5,003,080; 5,097,045; 5,103,024; 5,124,482; 5,149,837; 5,248,793; 5,280,126; 5,342,952, which are herein incorporated by reference, describe various processes and key intermediates for preparing atorvastatin.

Atorvastatin calcium produced by the processes described in the above mentioned United States patents does not give amorphous atorvastatin consistently but gives a mixture of its crystalline and amorphous forms, which has unsuitable filtration and drying characteristics and are not suitable for large-scale production.

PCT application, WO 97/03959, discloses novel crystalline forms of atorvastatin calcium designated as Form I, Form II, and Form IV and method for their preparation which provide more favourable filtration and drying characteristics.

PCT application WO 97/03960 describes a procedure for converting the crystalline form of atorvastatin to the amorphous form. Process disclosed therein comprises dissolving crystalline form-I atorvastatin in a non-hydroxylic solvent like tetrahydrofuran or mixtures of tetrahydrofuran and toluene. The process involves complete removal of the solvent under high temperature (about 90°C) and high vacuum (about 5mm) using capital intensive equipment. Exposure of the material to high temperature for several days leads to degradation of the product. This makes the process very inconvenient to operate at a large scale. Slow removal of solvents at a manufacturing scale renders this process as inefficient cost-wise and less productive.

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#### **SUMMARY OF THE INVENTION**

It is an objective of the present invention to provide an efficient process for the production of amorphous atorvastatin, which eliminates the problems of prior art and is convenient to operate on a commercial scale.

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Accordingly, the present invention provides a process for the preparation of atorvastatin calcium in an amorphous form which comprises dissolving crystalline atorvastatin in a non-hydroxylic solvent, adding a suitable non-polar hydro-carbon solvent and recovering atorvastatin from a solution thereof, by solvent precipitation, isolating and drying the product.

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Generally, the product can be isolated by any standard method known in the art such as by filtration, centrifugation or decantation. Typically, this product is isolated by filtration when any of the solvents within the scope of the process are used.

Major advantages of the present invention compared to the prior art processes are:

- i. elimination of the need to remove solvent by drying techniques.
- ii. less time consuming with improved filtration.
- iii. easy to operate on large-scale.
- iv. reproducibly produces amorphous product having allowable levels of residual solvents.

The present invention thus provides a novel process for the preparation of amorphous atorvastatin calcium and hydrates thereof which comprises:

- (a) dissolving crystalline atorvastatin calcium in a non-hydroxylic solvent;
- (b) adding a non-polar hydrocarbon anti-solvent to precipitate out the material; and
- (c) removing the solvent by filtration to afford amorphous atorvastatin calcium

The non-hydroxylic solvent is selected from a group of solvents, which have the ability to dissolve crystalline atorvastatin and includes tetrahydrofuran. Suitable non-polar hydrocarbon solvents are selected from a group consisting of: n-hexane, n-heptane, cyclohexane, hexane fraction, heptane fraction or the like. In a preferred embodiment of this invention, the non-hydroxylic solvent is tetrahydrofuran and anti-solvent is n-hexane, cyclohexane or n-heptane.

Generally, crystalline atorvastatin calcium is dissolved in a non-hydroxylic solvent, e.g. tetrahydrofuran, at a concentration of about 15% w/v to about 40% w/v, preferably at a concentration of about 25% w/v to about 15% w/v at ambient

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temperature and a non-polar hydrocarbon, preferably n-hexane, cyclohexane or n-heptane, is added at 0°C to 50°C, preferably at 20°C to 25°C. The product is recovered by filtration at ambient temperature. Filtration, which is fast and smooth, is carried out using nutsche filtration or centrifuge filtration. Preferably, nutsche filtration is used on large scale preparation. Filtered material, a semi-dry powder, is further dried to remove surface solvents in a vacuum tray drier, tray drier, fluid bed drier or a rotary vacuum drier to afford amorphous material. Preferably, material is dried in a vacuum tray drier at about 20°C to about 80°C for 6 hours to 24 hours. Most preferably, drying is carried out at about 50°C to about 60°C for 12 hours.

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Quantity of antisolvent varies from 5 times to 50 times the input of crystalline atorvastatin calcium depending upon its solution in non-hydroxylic solvent. Preferably, the quantity of antisolvent used is about 20 times to about 40 times the input of crystalline atorvastatin calcium to make overall concentration of about 5% w/v to about 2.5 w/v%.

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Amorphous atorvastatin calcium prepared according to the process of the present invention may be characterized by its x-ray powder diffration pattern (Figures 2) as shown in the accompanied drawings. X-ray powder diffration patterns (Figures 2) show no peaks which are characteristic of a crystalline atorvastatin calcium (Figure 1 of the accompanied drawings) thus demonstrating the amorphous nature of the product.

#### **BRIEF DESCRIPTION OF THE FIGURES**

Figure 1 is the diffractogram of crystalline atorvastatin calcium. The horizontal axis represents 20 and the vertical axis corresponds to peak intensity.

Figure 2 is diffractogram of amorphous atorvastatin calcium. The horizontal axis represents  $2\theta$  and the vertical axis corresponds to peak intensity.

The present invention is illustrated by the following examples, which are not intended to limit the effective scope of the claims.

## **DETAILED DESCRIPTION OF THE INVENTION**

### Example 1

[R-(R\*,R\*)-2-(4-fluorophenyl)- $\beta$ ,  $\delta$ -dihydroxy-5-(1-methylethyl)-3-phenyl-4-[(phenylamino)carbonyl)-1H-pyrrole-1-heptanoic acid hemicalcium salt (Amorphous Atorvasatin calcium).

#### Method A

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Crystalline atorvastatin calcium (10 kg) was dissolved in tetrahydrofuran (30 lt) under stirring at ambient temperature. Clear solution so obtained was added slowly to cyclohexane (350 lt) under nitrogen atmosphere. It was vigorously stirred maintaining temperature at 20-25°C. The precipitated product was centrifuged and dried under vacuum at about 60°C for 12 hours. Atorvastatin (9.5 kg) in an

amorphous form was obtained having residual solvent levels of 0.01% w/w tetrahydrofuran and 0.6% w/w cyclohexane. X-ray powder diffraction pattern (Figure 2 as shown in the accompanied drawings) demonstrate the amorphous nature of the product.

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#### Method B

Crystalline atorvastatin calcium (10 kg) was dissolved in tetrahydrofuran (30 lt) under stirring at ambient temperature. To a clear solution of atorvastatin, cyclohexane (350 lt) was added under vigorous stirring at 20 to 25°C. The precipitated mass was further stirred for 30 minutes and filtered in a centrifuge. The product was dried under vacuum at about 60°C for 12 hours. Atorvastatin (9.6 kg) in an amorphous form was obtained having residual solvent levels of 0.01% w/w for tetrahydrofuran and 0.7% w/w for cyclohexane. X-ray powder diffraction pattern demonstrates the amorphous nature of the product.

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#### Example 2

 $[R-(R^*,R^*)-2-(4-fluorophenyl)\beta, \qquad \delta-dihydroxy-5-(1-methylethyl)-3-phenyl-4-\\ [(phenylamino)carbonyl)-1H-pyrrole-1-heptanoic acid hemicalcium salt (Amorphous Atorvasatin calcium)$ 

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The process of Example 1 was repeated with crystalline atorvastatin calcium (10 kg) dissolved in tetrahydrofuran (30 lt) and using n-hexane instead of cyclohexane

to give amorphous atorvastatin (9.5 kg.). X-ray crystallography confirmed the amorphous nature of the product.

## Example 3

5 [R-(R\*,R\*)]-2-4-fluorophenyl)-β, δ-dihydroxy-5-(1-methylethyl)-3-phenyl-4[(phenylamino)carbonyl)-1H-pyrrole-1-heptanoic acid hemicalcium salt
(Amorphous Atorvasatin calcium)

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The process of Example 1 was repeated with crystalline atorvastatin calcium (10 kg) dissolved in tetrahydrofuran (30 lt) and using n-heptane instead of cylcohexane to give amorphous atorvastatin (9.6 kg). X-ray crystallography examination confirmed the amorphous nature of the product.

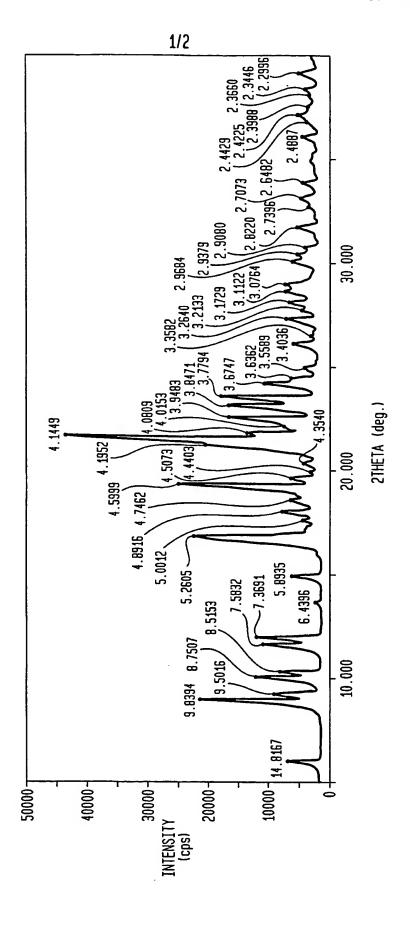
While the present invention has been described in terms of its specific embodiments, certain modifications and equivalents will be apparent to those skilled in the art and are intended to be included within the scope of the present invention.

#### WE CLAIM:

 A process for the preparation of amorphous atorvastatin calcium and hydrates thereof which comprises:

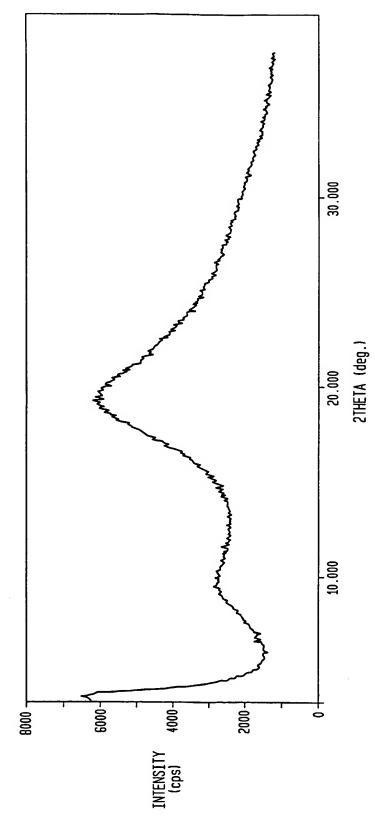
- (a) dissolving crystalline atorvastatin calcium in a non-hydroxylic solvent;
- (b) adding a non-polar hydrocarbon anti-solvent or adding the dissolved atorvastatin to the non-polar anti-solvent to precipitate out atorvastatin calcium; and
- (c) removing the solvent by filtration to afford amorphous atorvastatin calcium.
- 2. The process of claim 1, wherein the non-hydroxylic solvent is tetrahydrofuran and anti-solvent is chosen from a group of non-polar hydrocarbon solvents comprising n-hexane, cyclohexane or n-heptane.
- 3. The process of claim 1, wherein the non-hydroxylic solvent is tetrahydrofuran and anti-solvent is n-hexane.
- 4. The process of claim 1, wherein the non-hydroxylic solvent is tetrahydrofuran and anti-solvent is cylcohexane.
- 5. The process of claim 1, wherein the non-hydroxylic solvent is tetrahydrofuran and anti-solvent is n-heptane.
- 6. The process of claim 1, wherein said amorphous atorvastatin calcium is isolated by filtration.





WO 00/71116





SUBSTITUTE SHEET (RULE 26)

Interr nal Application No

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CLASSIF	A61K31/40 C07D207/34		
	International Patent Classification (IPC) or to both national class SEARCHED	sification and IPC	
	cumentation searched (classification system followed by classification	ication symbols)	
IPC 7	CO7D A61K		
Occumentati	on searched other than minimum documentation to the extent the	nat such documents are include	ed in the fields searched
Electronic da	ata base consulted during the international search (name of date	a base and, where practical, s	earch terms used)
C. DOCUM	ENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the	e relevant passages	Relevant to claim No.
A	WO 97 03959 A (WARNER LAMBERT C CHRISTOPHER A (US); JENNINGS R 6 February 1997 (1997-02-06)		1-6
	cited in the application		
A	DE 33 27 449 A (GLAXO GROUP LT 2 February 1984 (1984-02-02) page 9, line 7 - line 16 page 12, line 10 -page 13, lin		1-6
			1.0
Α	WO 97 03960 A (WARNER LAMBERT (US); SCHWEISS DIETER (US)) 6 February 1997 (1997-02-06) cited in the application	CO ;LIN MIN	1-6
	claim 1		
		-/	
			1
X Fur	ther documents are listed in the continuation of box C.	χ Patent family m	nembers are listed in annex.
"A" docum	ategories of cited documents: sent defining the general state of the art which is not dered to be of particular relevance	or priority date and	shed after the international filing date not in conflict with the application but the principle or theory underlying the
filing		"X" document of particul cannot be consider	ar relevance; the claimed invention ed novel or cannot be considered to
which citation	ent which may throw doubts on priority claim(s) or a is cited to establish the publication date of another on or other special reason (as specified) nent referring to an oral disclosure, use, exhibition or	"Y" document of particul cannot be consider	e step when the document is taken alone ar nelevance; the claimed invention ed to involve an inventive step when the ned with one or more other such docu-
other	means	ments, such combi in the art.	nation being obvious to a person skilled

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information on patent family members

Inter nal Application No PCT/IB 00/00014

					PC1/1B 00/00014	
Patent document cited in search report		Publication date		atent family member(s)	Publication date	
WO 9703959	A	06-02-1997	AU BG	6484296 A 102187 A	18-02-1997 30-10-1998	
			BR CA	9609872 A 2220018 A	23-03-1999 06-02-1997	
			CN	1190955 A	19-08-1998	
·		•	CZ	9800121 A	14-10-1998	
			ĔP	0848705 A	24-06-1998	
			HR	960339 A	30-04-1998	
			HU	9900678 A	28-07-1999	
			IL	122118 A	14-07-1999	
			JP	11509230 T	17-08-1999	
			NO Pl	980207 A 324496 A	16-01-1998 25-05-1998	
			SK	6298 A	07-10-1998	
			ÜS	5969156 A	19-10-1999	
DE 3327449	A	02-02-1984	AT	382154 B	26-01-1987	
			AT	276783 A	15-06-1986	
			AU AU	566881 B 1741783 A	05-11-1987 02-02-1984	
			BE	897422 A	30-01-1984	
			CA	1240313 A	09-08-1988	
			CH	657134 A	15-08-1986	
			CZ	280528 B	14-02-1996	
			CS	8305687 A	15-03-1988	
			CY	1434 A	02-09-1988	
			DE	3374010 D	12-11-1987	
			DK DK	68392 A	25-05-1992 31-01-1984	
			EP	349083 A,B, 0107276 A	02-05-1984	
			ES.	524590 D	01-06-1985	
			ËŠ	8505689 A	01-10-1985	
			FI	832757 A,B,	31-01-1984	
			FR	2531087 A	03-02-1984	
			GB	2127401 A,B	11-04-1984	
			GR	79349 A 84288 A	22-10-1984	
			HK HU	190603 B	28-10-1988 29-09-1986	
			IE	55748 B	02-01-1991	
			ΪĹ	69375 A	31-12-1986	
			IT	1168206 B	20-05-1987	
			JP	2025666 C	26-02-1996	
			JP	7030084 B	05-04-1995	
			JP KE	59044391 A	12-03-1984 03-06-1988	
			KE KR	3805 A 9100046 B	19-01-1991	
			LU	84935 A	23-11-1983	
			MY	5887 A	31-12-1987	
			NL	8302705 A	16-02-1984	
			NO	832773 A,B,	31-01-1984	
			NZ	205083 A	14-03-1986	
			PL	243228 A	27-08-1984	
			PT	77135 A,B	01-08-1983	
			SE SE	453195 B 8304208 A	18 <b>-</b> 01-1988 31-01-1984	
	•		SE SG	26088 G	31-01-1984 15-07-1988	
			JU	FAAAA A	72 A1 T200	
			SI	8311558 A	31-12-1995	

information on patent family members

Inten nal Application No PCT/IB 00/00014

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
DE 3327449	Α	<del>1</del>	SU	1266471 A	23-10-1986
			US	4994567 A	19-02-1991
			US	5013833 A	07-05-1991
			US	4562181 A	31-12-1985
			US	4820833 A	11-04-1989
			YU	155883 A	30-04-1986
			ZA	8305579 A	26-09-1984
·			ZW	17383 A	26-10-1983
WO 9703960	Α	06-02-1997	AU	700794 B	14-01-1999
			AU	6497896 A	18-02-1997
			BG	102188 A	31-08-1998
			BR	9609714 A	23-02-1999
			CA	2220455 A	06-02-1997
			CN	1190956 A	19-08-1998
			CZ	9800122 A	16-12-1998
			EP	0839132 A	06-05-1998
			HR	960312 A	28-02-1998
			IL	122161 A	14-07-1999
			JP	11510486 T	14-09-1999
			NO	980209 A	16-01-1998
			PL	324463 A	25-05-1998
			SK	5898 A	05-08-1998
WO 9703958	A	06-02-1997	AU	6484196 A	18-02-1997
			BG	102186 A	30-10-1998
			BR	9610567 A	06-07-1999
			CA	2220458 A	06-02-1997
			CN	1190957 A	19-08-1998
			CZ	9800123 A	17-06-1998
		•	EP	0848704 A	24-06-1998
			HR	960313 A	30-04-1998
			HN	9901687 A	28-10-1999
			IL	122162 A	14-07-1999
			JP	11509229 T	17-08-1999
			NO	980208 A	16-01-1998
			PL SK	324532 A	08-06-1998 06-05-1998
			3K	5998 A	00-00-1990